

Study Plan for the U.S. Coast Guard Survey of Lake Erie in Winter 2018-19

1. Project Objective

The biogeochemical processes of Lake Erie during the winter are relatively unknown and represent an important uncertainty in our understanding of this Great Lake. U.S. Coast Guard operations on the Lake during the winter offer a valuable opportunity for data collection to fill this gap in our knowledge. This project uses current USCG operations as a sampling platform to measure the distribution of phytoplankton biomass and dissolved nutrients through Lake Erie in the winter.

2. Project Design

The project consists of synoptic sampling of the near-surface waters of Lake Erie during normal operations of the USGC *NEAH BAY*. The cutter *NEAH BAY* operates throughout Lake Erie during the winter season and offers an unparalleled platform for sampling. With the support and leadership of LCDR Andy Perodeau, Commanding Officer of *NEAH BAY*, water samples will be collected when mission permits to provide a spatial and temporal survey of the Lake to identify and enumerate phytoplankton, determine the concentrations of particulate chlorophyll *a* and dissolved and particulate nutrients.

The varied and unpredictable nature of USCG operations on Lake Erie in the winter season necessitates a flexible sampling strategy to maximize the spatial and temporal coverage of the survey.

3. Project Parameters

a) Sampling location, time, and local conditions

The latitude, longitude (decimal degree format with 4 decimal minimum) and time will be recorded from the ship's navigational suite at every sampling location. Local environmental conditions, including air and water temperature, wind direction and strength, cloud cover, ice cover, and ice thickness will also be recorded.

*b) Particulate chlorophyll *a* concentration*

Near-surface water (1 m depth) will be collected using a stainless steel sampling bottle and processed using the chlorophyll *a* standard operating procedure (Appendix A). We have adopted this approach rather than use of Go-Flo bottles to accommodate working in ice. These bottles were custom made (welded stainless steel) by Fletcher Manufacturing, Bowling Green, OH).

c) Cell identification and enumeration

Near-surface water (1 m depth) will be collected using a stainless steel sampling bottle. Aliquots are removed to 50 mL polyethylene Falcon tubes and preserved with several drops of Lugol's iodine solution. Microscopic analysis is performed by

William Cody, Aquatic Taxonomy Specialists, Malinta, OH.

d) Dissolved and particulate nutrient concentrations

Near-surface water (1 m depth) will be collected using a stainless steel sampling bottle and stored frozen until analysis at Heidelberg University (National Center for Water Quality Research). Heidelberg follows established U.S. Environmental Protection Agency methods for analysis of nutrients and major ions (Appendix B).

4. Sampling Design

- a. The ship will come to a stop at the sampling station, chosen at regular time intervals or by the discretion of the ship's command. Stations will not be biased on the basis of ice conditions or location. The location and time of the sampling station will be recorded along with the environmental conditions.
- b. Designated crew members of *NEAH BAY*, trained in sampling and supervised by LCDR Perodeau, will deploy the stainless steel sampling bottle to a depth of 1 m and collect the water grab sample.
- c. Water samples are transferred to 1 L opaque acid-cleaned polyethylene storage bottles and stored at 4 °C until picked up by personnel by BGSU (same day). Samples are transported to BGSU in cooler containing ice packs.
- d. Sub-samples for particulate chlorophyll *a* and dissolved nutrient concentrations will be processed by BGSU personnel according to the Standard Operating Procedures and stored under the appropriate conditions until shipping and/or analysis.
- e. Particulate chlorophyll *a* samples will be processed according to the EPA standard 445.0 method for chlorophyll *a* analysis. Analysis will be completed prior to the 3.5 week hold time allowed under this procedure.

For analysis by fluorometry, samples are extracted in 90% acetone (24 h at -20° C) with sonication and chlorophyll measured in a TD-700 fluorometer (Turner) using the non-acidified approach (Welschmeyer, 1994)

- f. Dissolved nutrient samples will be shipped to the National Center for Water Quality Research at Heidelberg University (Tiffin, OH) for analysis. Analytical methods used by Heidelberg are described in Appendix B.

5. Qualified Personnel

Project Manager

Ohio EPA Level 3 Certification: Chemical Water Quality Assessment

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6. References

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Appendix A

Standard Operating Procedure for Chlorophyll-*a* Sampling Method: Field Procedure for Use by U.S. Coast Guard

1.0 Scope and Application

This method is used to filter chlorophyll-*a* samples from the Great Lakes and Tributary streams.

2.0 Summary of Method

A grab lake water sample is collected from a stainless steel sampling bottle at various depths and filtered by vacuum filtration in dim light. The filter is then placed in a screw cap polyethylene culture tube in the dark. The tube is stored in the dark at sub-freezing temperatures prior to extraction and analysis. The BGSU laboratory will follow protocol LG405, developed by the EPA's Great Lakes National Program Office (GLNPO) for water quality surveys of the Great Lakes (appended).

3.0 Apparatus

Plastic filter funnel, Pall Filtron (250 mL capacity)
Vacuum manifold system to accommodate 3 filter funnels
Vacuum system (3-4 psi)
GF/F filters, Whatman (25 mm)
Screw cap polyethylene tubes
Graduated polystyrene pipettes (25 mL; disposable)
Pasteur short disposable pipets
Rubber bulb
Plastic wash bottle, 500 mL
Plastic wash bottle, 500 mL, for MgCO₃
Filter forceps
Opaque sample bottles, 1000 mL (Nalgene or equivalent)

4.0 Reagents

Saturated Magnesium Carbonate Solution Add 10 grams magnesium carbonate to 1000 mL of deionized water. The solution is settled for a minimum of 48 hours. Decant the clear solution into a new container for subsequent use. *Only the clear "powder free" solution is used during subsequent steps.*

5.0 Sample Handling and Preservation

Sample collection and preservation will follow the procedures described in the Manual of Ohio EPA Surveillance Methods and Quality Assurance Practices (Ohio EPA, 2009) and the Inland Lakes Sampling Procedure Manual (Ohio EPA, 2010). The entire procedure should be carried out as much as is possible in subdued light to prevent photodecomposition. The frozen samples should also be protected from light during storage for the same reason. During the filtration process, the samples are treated with MgCO₃ solution (section 4) to eliminate acid induced transformation of chlorophyll to its degradation product, pheophytin. Samples are stored by station in aluminum foil and transported to the BGSU laboratory in a cooler with dry ice. Analysis should be performed as soon as possible following sampling.

6.0 Field Procedure

- 6.1 Following sample collection with the stainless steel sampling bottle, samples are transferred to 1000 mL opaque Nalgene bottles, labeled with the station, sample depth, *eg.* Surface, representing a surface sample
- 6.2 Place filters, using forceps, textured side up. Assemble the filtration apparatus just prior to filtration.
- 6.3 Due to differing trophic levels among the Great Lakes, the volume of water filtered varies. For

Lake Erie, 25 mLs of sample are filtered. After inverting the sample bottle several times to create a uniform mixture, carefully draw 25 mL into a pipette and distribute contents into filtration funnel.

6.4 Turn vacuum pressure on, not exceeding 3 psi. Our plans call for use of a hand pump.

Check Frequently During Filtration to Insure Pressure Does Not Go Above 3 PSI!!!

6.5 When approximately 10 mL of sample remains on the filter, add 10 drops of the MgCO_3 (section 4.1) solution using a disposable pipet. Thoroughly rinse the filter apparatus and graduated cylinder, using a squirt bottle, with deionized water. Turn off vacuum pressure as soon as the liquid disappears to prevent the breakage of cells.

6.6 Using the forceps, fold and remove the filter and carefully place it into the bottom portion of the prelabeled culture tube (see section 10) and close tightly. Lay all tubes flat and completely wrap in aluminum foil. Clearly label the Lake, station and date on masking tape and attach to above mentioned aluminum foil package. Immediately freeze. All the above procedures should be completed in subdued light.

7.0 Quality Control

7.1 Each of the following audits is collected once per lake transect.

7.2 Field duplicates are taken from a second stainless steel sampling bottle collected at about the same time and location as the regular field sample. It is transported from the Niskin bottle to the onboard biology laboratory in an opaque bottle marked as duplicate sample.

7.3 Laboratory duplicates are filtered from the same opaque sample bottle as their corresponding regular field samples.

7.4 Field blanks, consisting of reagent water are carried by an opaque sample bottle from the onboard reagent water supply to the filtration apparatus. The bottle is used only for field blanks and is permanently marked as such.

8.0 Waste Disposal

Follow all laboratory waste disposal guidelines regarding the disposal of MgCO_3 solutions.

9.0 Shipping

Once a transect has been completed or a batch of 35 samples has been completed, wrap all samples into one complete batch and clearly label with date. Pack tightly in a medium sized cooler and fill all spaces with enough dry ice to last 24 hours. Dry ice is considered a hazardous chemical by most shipping companies and has to be accompanied by authorizing paperwork. Once transported to BGSU, the samples should be immediately placed in the freezer.

10.0 Labeling

Sample identification information is provided on printed labels both prior to and during the survey. The labels are affixed to the side of the 16×100 mm chlorophyll tube. The sample identification number is covered with clear tape in case the tube becomes wet.